



**DF-1519**

**M. Sc. (Sem. III) Examination**  
**March/April – 2016**  
**Physics : SPL : Materials Science**  
*[PH-(M) 534 Advanced Materials Science]*

Time : 3 Hours]

[Total Marks : 70

**Instructions :**

(1)

नीचे दशांशवैध निशानियाणी विगतो उत्तरवडी पर अवश्य लपवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="text" value="M. SC. (SEM. III)"/>	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="text" value="PHYSICS : SPL : MATERIALS SCIENCE"/>	<input type="text"/>
Subject Code No. : <input type="text" value="1"/> <input type="text" value="5"/> <input type="text" value="1"/> <input type="text" value="9"/>	<input type="text"/>
Section No. (1, 2,.....): <input type="text" value="Nil"/>	
Student's Signature	

- (2) Figures to the right hand side indicate full marks.
- (3) Assume data if required.
- (4) Symbols have their usual meaning.
- (5) Non scientific calculator is permissible to use.

1 Attempt any two :

- (a) Explain the following terms : 7
  - (i) Biomaterials
  - (ii) Spintronic
  - (iii) Fullerene.
- (b) (i) Write a brief note on Materials and Civilization 4  
with suitable examples.
  - (ii) Explain schematic diagram of Materials science. 3
- (c) (i) Explain a piece of wood as a material. Describe 3  
some of its structure characteristics in light of materials science.
  - (ii) How ceramics and composites are differ ? Explain 4  
with suitable examples.

- 2** Attempt any two :
- (a) (i) Briefly explain “Theory of Liquids” and list their findings. **3**
- (ii) How transitions between state of matter can takes place ? Justify relevant thermodynamical parameters. **4**
- (b) Explain the following terms : **7**
- (i) Glass
- (ii) Emulsion
- (iii) Liquid crystals
- (c) State the following phases and give exemples of such states and their significance in Materials Science. **7**
- |            |            |
|------------|------------|
| Dispersion | Dispersive |
| State      | State      |
| Solid      | Solid      |
| Solid      | Liquid     |
| Solid      | Gas        |
| Liquid     | Solid      |
| Liquid     | Liquid     |
| Liquid     | Gas        |
| Gas        | Solid      |
| Gas        | Liquid     |
- 3** Attempt any two : **7**
- (a) Explain the bonds presents in Macro molecules. State role of secondary bonds in macro molecules with suitable examples.
- (b) (i) “Protein, Nucleic Acid, DNA and RNA are macro molecules, it is possible to determine their structures through development in Materials Science”. Justify this statement. **4**
- (ii) Differentiate  $r$  and  $R$  in macro molecules. Show the effect of  $r/R$  ratio on their three dimensional structure co-ordination, number. **3**
- (c) (i) Explain Hydrogen Bridge in detail. **3**
- (ii) Find the critical radius ratio for triangular coordination. **4**

- 4 Attempt any two : 7
- (a) What do we call the system in which complete liquid and solid solubility of two component is observed ? Take the example of Copper-Nickel system. Draw its phase diagram and explain it.
- (b) (i) State Gibb's phase rule. Explain all terms with suitable examples. 4
- (ii) Classify non-crystalline phases. Explain properties of phases in any one non crystalline phase. 3
- (c) (i) What informations we avail from the phase diagram ? State the significance of phase diagrams in materials science. 4
- (ii) Give the statement of Lever's rule and list the applications of phase diagrams. 3
- 5 Attempt any two :
- (a) Wood, Teeth and Horn may in which class of materials ? Explain such class of materials in detail. 7
- (b) How the non equilibrium conditions may explain in phase diagram ? Briefly explain such conditions with suitable examples. 7
- (c) (i) List the types of ceramics and their uses. 4
- (ii) Hemicellulose ( $C_6H_{10}O_5$ ) is an important component in wood. Calculate its molecular weight for a degree of polymerization of 150. 3
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